Bio-oxidative Cleavage of Carotenoids: Important Route to Physiological Active Plant Constituents

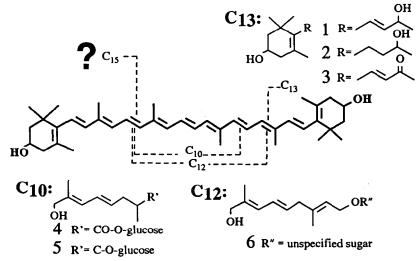
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Abstract: The β -D-glucosides of $(4R,1^{*}E,3^{*}E)-4-(5^{*}-hydroxy-3^{*}-methyl-1^{*},3^{*}-pentadienyl)-$ 3,5,5-trimethyl-2-cyclohexen-1-one 7 as well as of*trans*-abscisic alcohol (ABA-alcohol) 8have been isolated and characterized in quince fruit through spectral and chemical studies.ABA-alcohol 8 has recently been found to be involved in the biosynthesis of the importantplant hormone abscisic acid (ABA) 9, the latter compound being also present in a stillnon-specified glycosidically bound form in quince fruit. Based on the finding of these newfifteen carbon (C₁₅) constituents in quince, biodegradation of carotenoids is discussed.

It has previously been shown by our group that quince (Cydonia oblonga Mill.) fruit contains a great number of C_{13} -norisoprenoid compounds, such as, e.g., ionone structures <u>1-3</u>, most of them playing an important role as flavour precursors^{1,2}. The additional finding of the irregular terpenoids <u>4-6</u>³⁻⁵, which are apparently derived from the central part of the carotenoid chain, suggests a hypothetical cleavage of quince carotenoids as schematically outlined below:

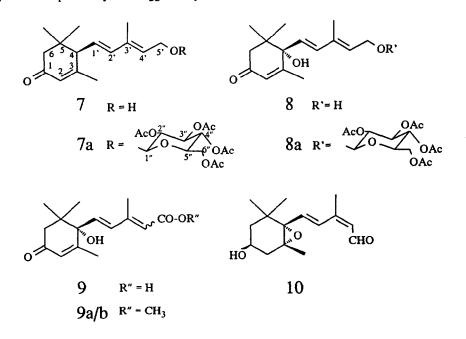


On the basis of the above shown assumed cleavage of quince carotenoids, C_{15} -constituents would be further logical carotenoid metabolites in the fruit. In an effort to isolate these C_{15} -carotenoid end groups, polar extracts of quince juice have been prepared by the following methods:

adsorption of glycosidic constituents on Amberlite XAD-2 resin according to the method of Günata *et al.*⁶, followed by ethyl acetate elution;

(ii) isolation of non-glycosidic constituents using continuous diethyl ether extraction.

Eluates obtained by method (i) were further fractionated with the aid of rotation locular countercurrent chromatography (RLCC)⁷. After acetylation and flash chromatography⁸ a final purification was achieved by preparative HPLC on LiChrospher Si 100 (eluent: Et₂O). In this way, the B-D-glucopyranoside of ketoalcohol 7 has recently been identified by us as the first C15-structure in quince fruit. More detailed information about the identification of the new natural product 7a including all spectral data have been published elsewhere⁹. Most recent investigations revealed now the presence of the B-D-glucopyranoside of ABA-alcohol as further C15-structure in quince¹⁰. In addition to glucoside 8a non-conjugated ABA-alcohol 8 was isolated from quince extracts prepared by method (ii)9. Whereas glucoside 8a was identified by us for the first time in nature, aglycone 8 is well-known and has attracted considerable research interest due to its implication in the biosynthesis of the important plant hormone ABA $9^{11,12}$. In a recent study it has been shown that alcohol 8 is an intermediate in ABA-biosynthesis in a shunt pathway from ABA-aldehyde involving enzymatic reduction to ABA-alcohol 8 and oxidation of the latter compound to ABA 9 via a cytochrome P-450 monooxygenase¹¹. The finding of high amounts of glucoside 8a (approx. 6 mg per kg of fresh fruit) indicates that ABA-alcohol 8 accumulates in form of its B-D-glucopyranoside in the fruit. However, for quince fruit it is still not known if this accumulation of 8 may generate a precursor pool for subsequent ABA biosynthesis; this possible role of aglycone 8 has previously been suggested by Linforth et al.¹²



(i)

The detection of free and glycosidically bound ABA-alcohol **8/8a** prompted us to search also for the presence of the important plant hormone ABA **9** in quince. After enzymatic hydrolysis of separated RLCC fractions followed by methylation of the liberated aglycones with etheral diazomethane¹³, two isomers of ABA **9** - tentatively identified as the *cis,trans-* and *trans,trans-*derivatives **9a/b** by comparison with a commercially available reference (*Sigma*) - were identified by HRGC-MS- and HRGC-FTIR-analyses. The final structural elucidation of the conjugating sugar moiety is the subject of active research.

Concerning the biogeneration of the C_{15} -structures under investigation there have been two main approaches to explain their formation from carotenoids. Isoe¹⁴ initially proposed a variety of C_{15} - and C_{13} -degraded carotenoid structures to be derived by photooxidative cleavage of the polyene chain. In recent years, however, biodegradation processes have been favoured, assuming a hypothetical dioxygenase as carotenoid-degrading enzyme¹⁵. The thoroughly studied enzyme lipoxygenase [EC 1.13.11.12] is such a dioxygenase-type biocatalyst. In the so-called "co-oxidation" process in the presence of certain unsaturated fatty acids this enzyme was found to cleave carotenoids *in vitro* to C_{13} - and C_{15} -fragments^{16,17}. Lipoxygenase-catalyzed degradation, e.g., of the ubiquitous epoxy-carotenoid violaxanthin yielded xanthoxin 10 as initial C_{15} -cleavage product¹⁷. From xanthoxin 10, which - up to now - could not be detected by us in quince, the enzymatic pathways giving rise to the formation of ABA 9 have been extensively studied and much progress regarding ABA generation from precursor 10 has been made in very recent years¹⁸⁻²². Despite these efforts, almost nothing is known about the initial enzymatic cleavage of the carotenoid chain *in vivo*. And since - in addition to lipoxygenase - other carotenoid oxidizing catalytic systems are operative in plant tissues, further investigations are necessary to finally elucidate the important initial step of carotenoid degradation in plants.

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- 10. Spectral data of 8a: UV: 232 nm; ¹H NMR (400 MHz, CDCl₃): 1.00 and 1.08 (2s, 2 x 3H, 2 x CH₃-C(5)); 1.76, (br.s, 3H, CH₃-C(3')); 1.89 (d, 3H, J = 1.3, CH₃-C(3)); 1.99 2.07 (4s, 4 x 3H, 4 x CH₃-CO); 2.23 (d, 1H, J = 16.8, H_a-C(6)); 2.45 (d, 1H, J = 16.8, H_b-C(6)); 3.67 (dxdxd; 1H, J = 9.9, 4.7, 2.5, H-C(5'')); 4.14 (dxd, 1H, J = 12.3, 2.5, H_a-C(6''); 4.23 (dxd, 1H, J = 12.3, 4.7, H_b-C(6'')); 4.29 (dxd, 1H, J = 12.6, 7.4, H_a-C(5')); 4.41 (dxd; 1H, J = 12.6, 6.4, H_b-C(5')); 4.52 (d, 1H, J = 8.0, H-C(1''); 4.98 (dxd, 1H, J = 9.6, 8.0, H-C(2'')); 5.08 (dxd, 1H, J = 9.9, 9.5, H-C(4'')); 5.19 (dxd, 1H, J = 9.6, 9.5, H-C(3'')); 5.58 (br.t, 1H, J = 6.7, H-C(4')); 5.73 (d, 1H, J = 15.7, H-C(1')); 5.90 (br.s, 1H, H-C(2)); 6.32 (d, 1H, J = 15.7, H-C(2')). ¹³C NMR (50 MHz, CDCl₃): δ 12.8 (CH₃-C3'), 19.0 (CH₃-C3), 20.5-20.6 (CH₃-CO), 22.9 and 24.1 (2 x CH₃-C5), 41.5 (C5), 49.8 (C6), 62.0 (C6"), 65.3 (C5'), 66.5 (C4"), 71.3 (C2"), 71.9 (C5"), 72.8 (C3"), 79.5 (C4), 99.5 (C1"), 126.8 (C2), 127.5 (C4'), 128.8 (C1'), 134.4 (C2'), 136.5 (C3'), 162.9 (C3), 169.3-171.0 (4 x CH₃-CO), 197.8 (C1) (signals were assigned on the basis of two-dimensional ¹H, ¹³C-COSY-experiments); CD-spectra: identical with the published one⁹; thermospray-MS: 598 [M+NH₄]⁺.
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